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Visible-Light mediated Controlled and Living Radical Polymerization of Functional Vinyl Monomers by

N-substituted benzo[ghi]perylene derivatives as organocatalyst

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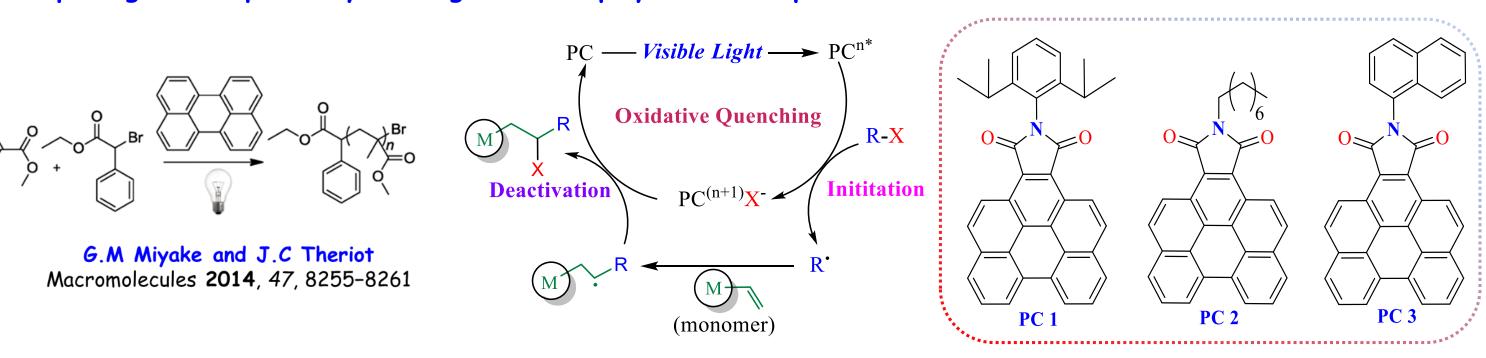


Abstract

Controlled radical polymerization is essential for synthesizing well-defined functional polymers. Recently, organic chromophores have enabled visible light-induced ATRP using alkyl bromide initiators. In this work, we explore Nsubstituted benzo[ghi]perylene (BPIm) as an efficient organic photocatalyst. Under visible light, BPIm mediates the controlled polymerization of various vinyl monomers (MMA, BMA, LMA, styrene, and NIPAM), yielding polymers with predictable molecular weights and low dispersity. Notably, the polymerization shows excellent temporal control, as it can be paused and resumed by toggling the light.

Aim of the study

- \Box To design and develop benzo[ghi]perylene-based chromophores with strong π - π * absorption in the visible region for efficient light harvesting in photoredox catalysis.
- ☐ To introduce N-substituents on the benzo[ghi]perylene core to enhance solubility and improve processability compared to the parent structures.
- □ Exploring its compatibility with light-driven polymerization processes.



Experimental section

General Polymerization procedure using PC [1-3] as Photocatalyst:

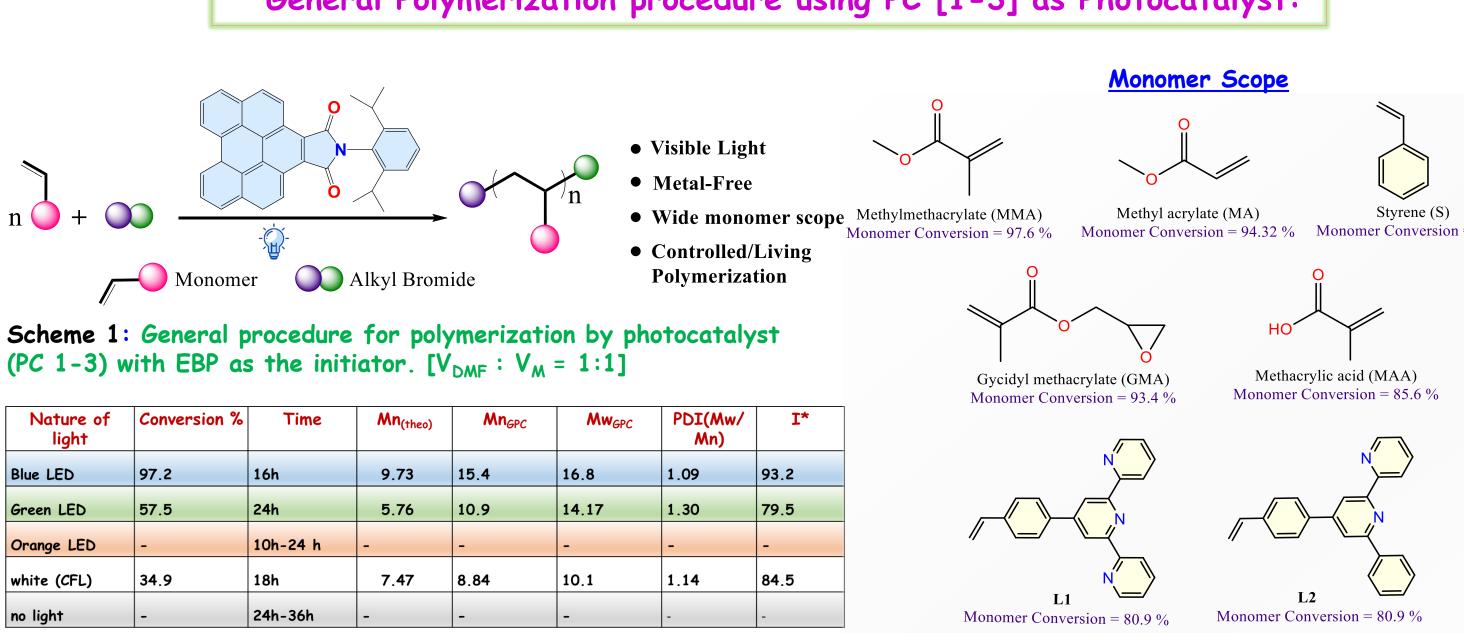
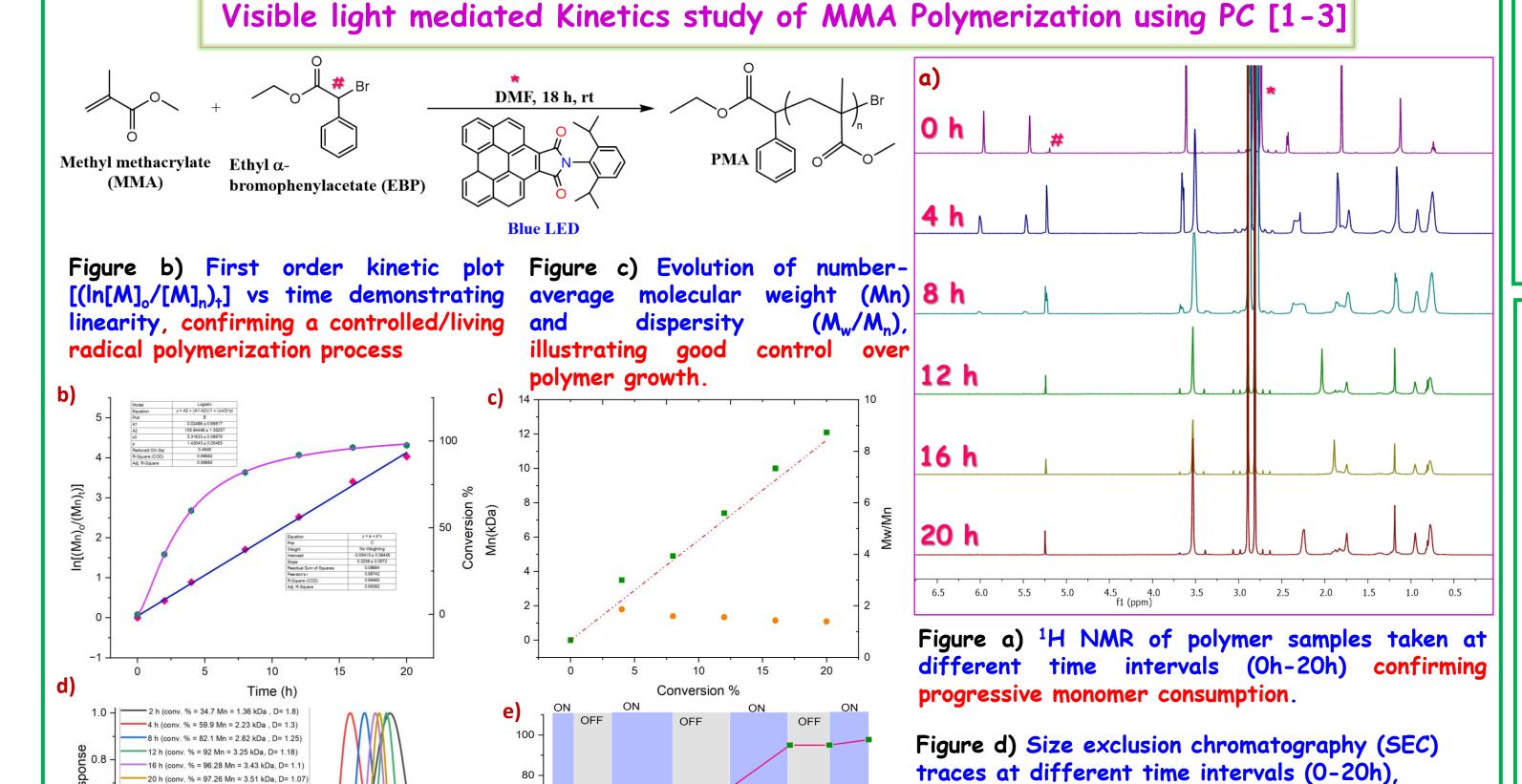


Table 1: Comparison of Photopolymerization under different light source by photocatalyst (PC 1) with EBP as the initiator. $[V_{DMF}: V_{MMA} = 1:1]$

- > PC 1 not only efficiently photopolymerizes traditional monomers but also vinyl-appended NNC-type ligands, highlighting its broad polymerization capability.
- > Moreover, it demonstrates compatibility with acid-sensitive monomers such as MAA and GMA



showing unimodal, symmetrical molecular weight

on/off

further

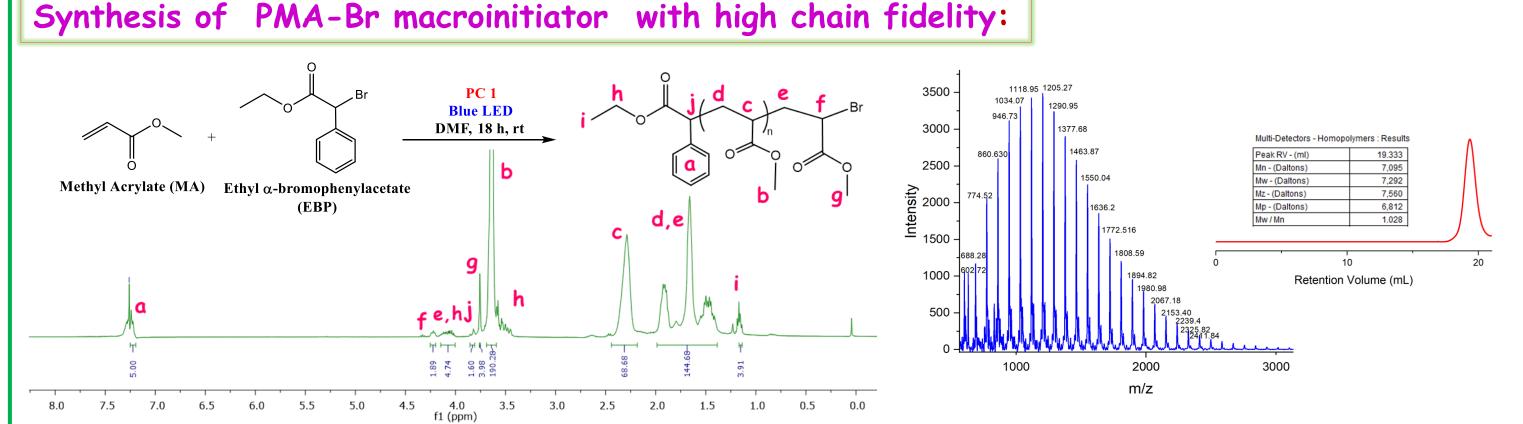
experiment showing

confirming the

distributions.

photocontrolled nature.

0 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 Retention Volume (1min/1ml) Time (h)



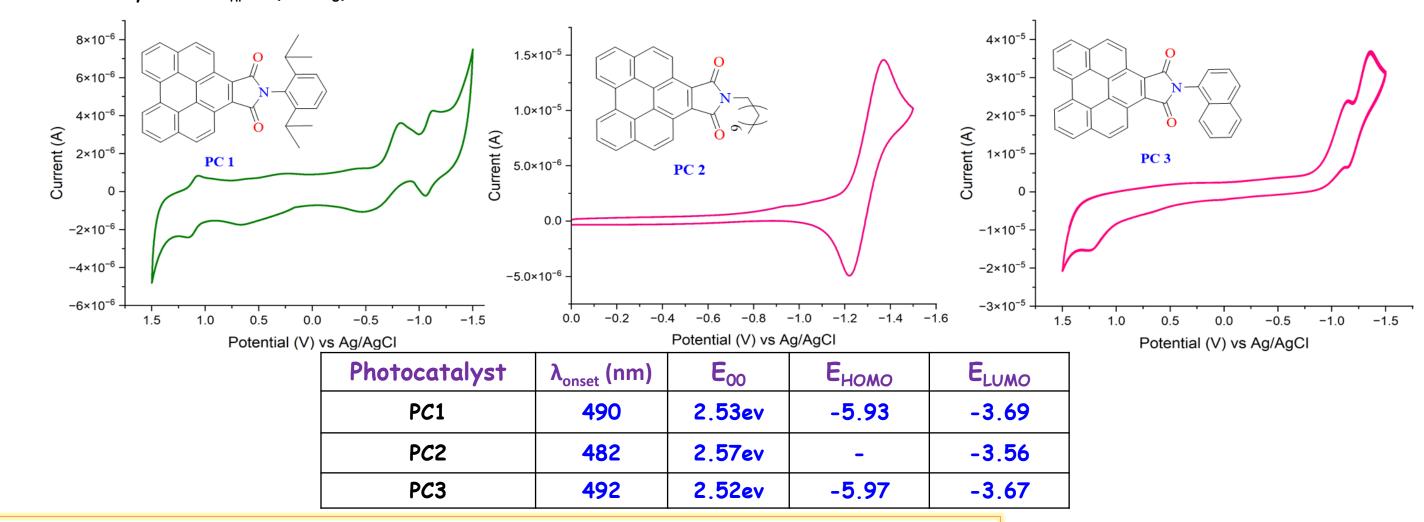
Results and Discussion

Photophysical & Electrochemical Properties of PC 1-3:

- □ PC1 is the most balanced: good light absorption, long-lived excited state, and reasonable radiative decay.
- □ PC2 is inefficient due to weak absorption and shorter excited-state lifetime. □ PC3 absorbs light best but loses energy too quickly non-radiatively

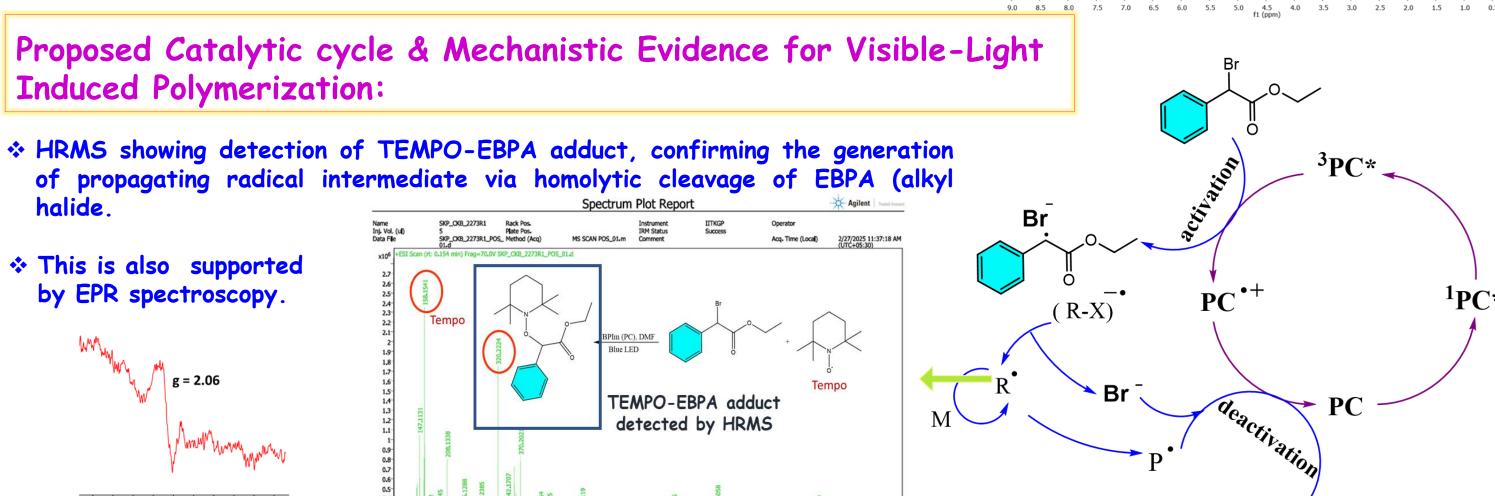
Photocatalyst	λ _{max} (nm) ^a	ε _{max} (10 ⁴ M ⁻¹ cm ⁻¹) ^b	λ _{em} (nm) ^c	Stokes shift (nm)	Фѕ	τ (ns)e	k _r (s ⁻¹) ^f	k _{nr} (s ⁻¹) ^f
PC1	338, 390, 482	5.2, 1.62,0 .8	529	191	0.11	7.85	1.50×10^7	1.21 × 10 ⁸
PC2	337, 388, 476	4.56, 1.52, 8.5	525	188	0.17	7.385	2.17×10^7	1.06 × 10 ⁸
PC3	339,389,477	8.2,2.5,1.1	531	192	0.076	6.97	1.09×10^7	1.33 × 10 ⁸

Table 1: a) λ_{max} (maximum absorption wavelength) measured in DMF by UV-vis absorption spectra. b) Molar absorptivity calculated at λ_{max} in DMF. c) Emission wavelength recorded by fluorescence spectroscopy in DMF (λ excitation =390) (d) Relative fluorescence quantum yield determination of PCs 1-3 using quinine sulphate ($\Phi=0.546$ in 0.1 M H₂SO₄) as the reference standard. e) Lifetime recorded in DMF by TCSPC f) rate constants are calculated using the following equations: Radiative decay rate: $k_r = \Phi_s / \tau$, Nonradiative decay rate: $k_{nr} = (1 - \Phi_s) / \tau$



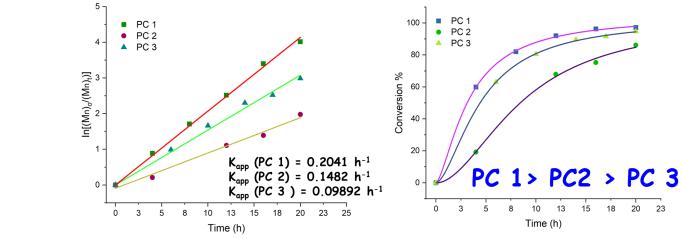
Sequence-controlled polymers study with N-heterocyclic ligands:

* Schematic represention of block copolymer synthesis by sequential addition of LMA and NNC demonstrating temporal control and living nature of polymerization 92% LMA 8% NNC



Conclusion

- 1. N-substituted benzo[ghi]perylene absorbs strongly in the visible region, particularly around 450 nm, aligning well with blue LED excitation (2.76 eV) due to its E_{00} of 2.53 eV, enabling efficient photocatalysis.
- 2. Its excellent visible-light compatibility and photostability make it well-suited for spatiotemporally controlled photopolymerization.
- 3. he N-substituted derivative shows a large Stokes shift (~192 nm), possibly due to ICT and excited-state relaxation, which may help suppress self-absorption and enhance PET.
- 4. Among the series, PC 1 delivers the highest polymerization yield and exhibits superior control over polymer growth compared to PC 2 and PC 3.



P-Br

References

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