





CATIONIC PHOTOPOLYMERIZATION OF BIOBASED OXETANE MONOMERS FROM ADIPIC, ITACONIC AND CITRIC ACID FUNCTIONALIZATION

Dumitru Moraru¹, Lazaros Papadopoulos², Minna Hakkarainen², Marco Sangermano¹

¹Politecnico di Torino, Italy ²KTH Royal Institute of Technology, Sweden

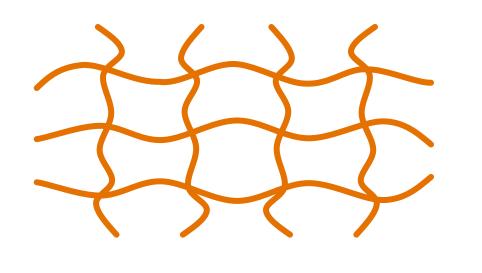
Introduction

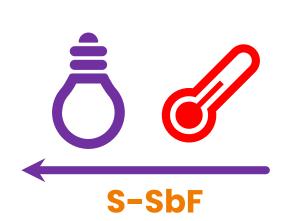
In the context of sustainable materials development, **biobased monomers** are essential for reducing reliance on fossil-based chemicals. This study reports the **synthesis of three novel oxetane monomers**, derived from the esters of adipic, itaconic, and citric acids (**BOA, BOI, and TOC**), via a one-pot transesterification reaction with 3-ethyl-3-oxetanemethanol.

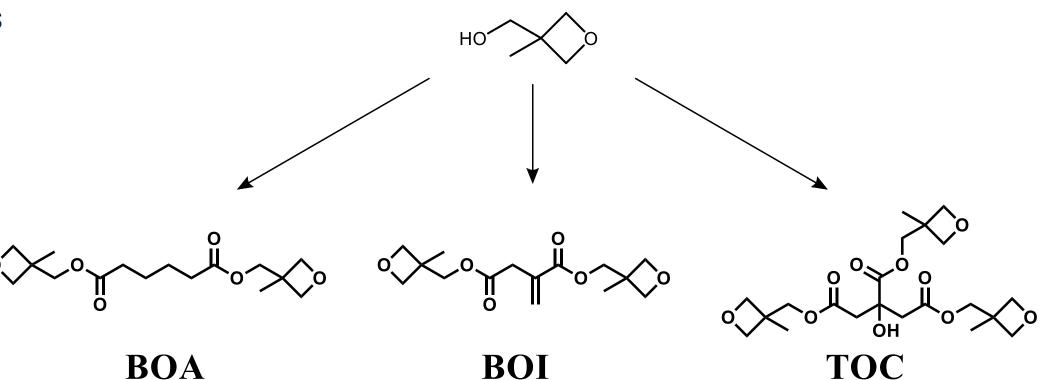
Cationic ring-opening photopolymerization (CROP) was employed to produce crosslinked materials. Characterization by FTIR, photo-DSC, DSC, and DMTA demonstrated effective curing kinetics and promising thermomechanical properties, particularly at irradiation temperatures above 85 °C. This work contributes to the development of sustainable, high-performance materials.

Synthesis and Photopolymerization

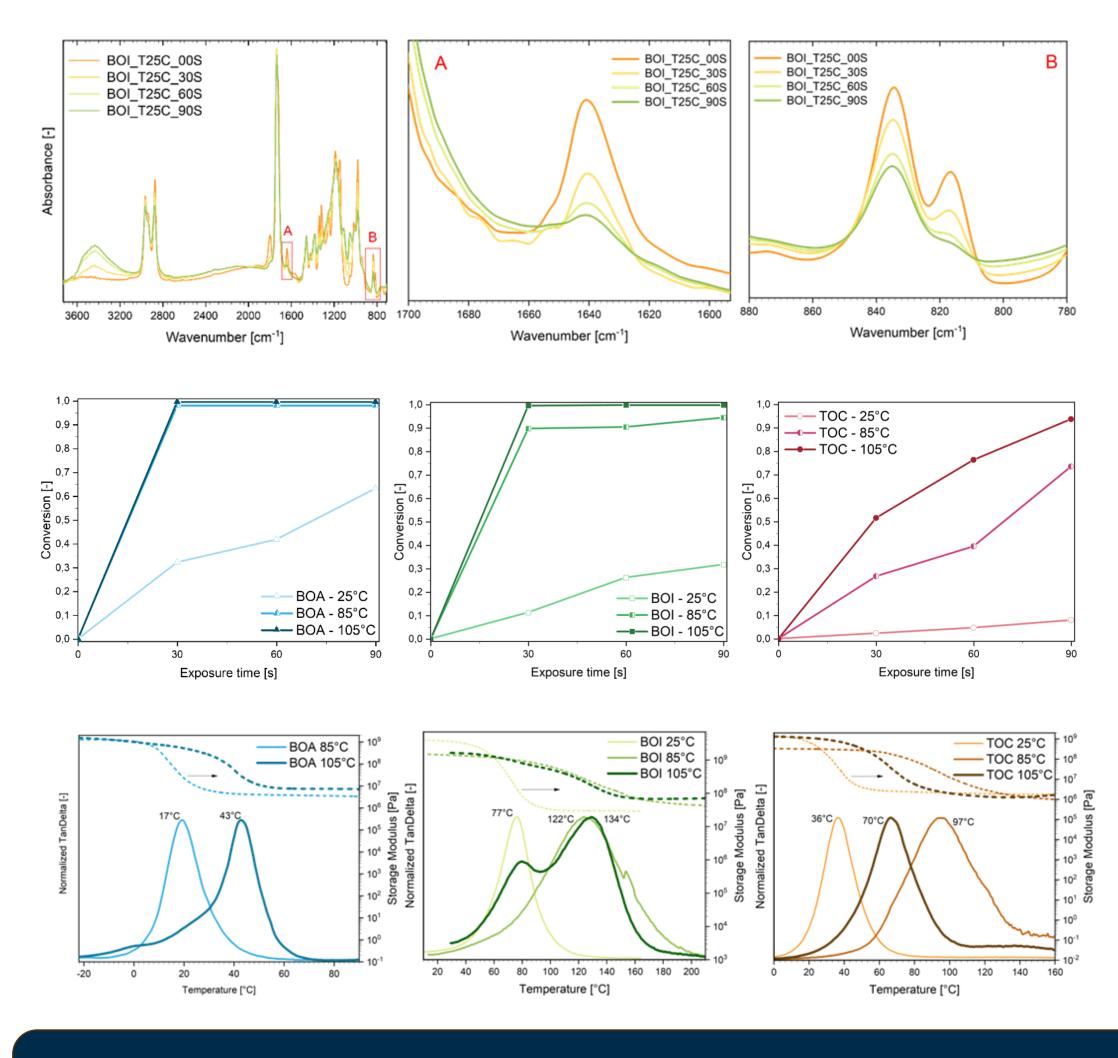
- **Transesterification** reaction using 3-ethyl-3-oxetane methanol with di-/tri-carboxylic esters (DMA, DMI, TEC), catalysed by DBTO.
- Photoinitiator: triarylsulfonium hexafluoroantimonate salts (S-SbF)



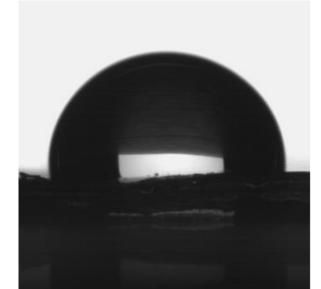


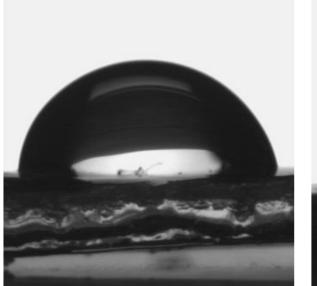


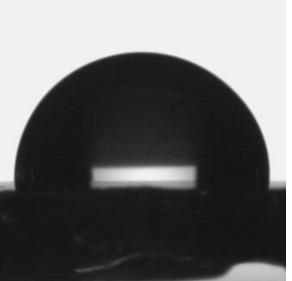
Results



Curing Temperature		DMTA Tg [°C]	Modulus [Pa]	Contact angle [°]	Gel content [%]
ВОА	25 °C	N/A	N/A	NA	NA
	85 °C	17	$3,69 \times 10^6$	93±4	89
	105 °C	43	$7,03 \times 10^6$	95±6	89
ВОІ	25 °C	77	$2,99 \times 10^7$	NA	NA
	85 °C	80/122	5,68 × 10 ⁷	72±4	100
	105 °C	134	$7,10 \times 10^7$	73±5	100
тос	25 °C	36	$1,47 \times 10^6$	NA	NA
	85 °C	97	$1,24 \times 10^6$	92±3	100
	105 °C	70	1,41 × 10 ⁶	94±2	100







Conclusions

- Bio-based oxetane monomers show promising reactivity in UV curing, especially BOI with dual reactivity.
- High oxetane group conversion (>85 °C curing).
- BOI exhibits highest Tg and crosslink density, suggesting best performance for applications demanding rigidity.
- TOC's hydroxyl groups lead to ether linkages, lowering crosslinking density, and Tg.

Acknowledgments

