Phthalate Extraction from Waste Blood Bag PVC using Supercritical CO₂ is Effective

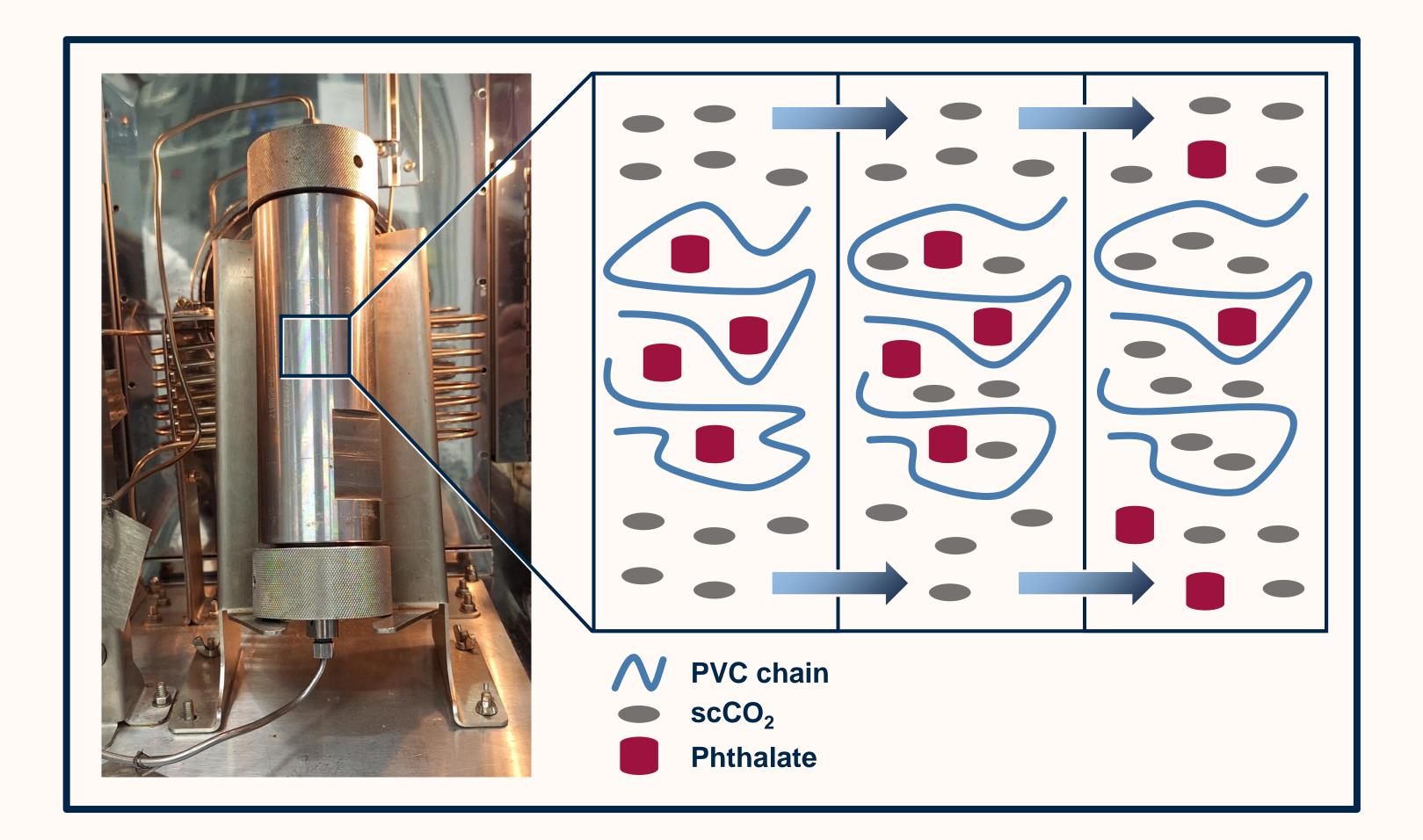
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Key Findings

- Influence of sample thickness > pressure > temperature
 - Grinding step potentially viable
- Extraction efficiency (EE) versus time: stretched exponential decay
- Only minor increase in EE above 110 °C, 200 bar, and 60 minutes
- ¹H-NMR works as a method for determining EE

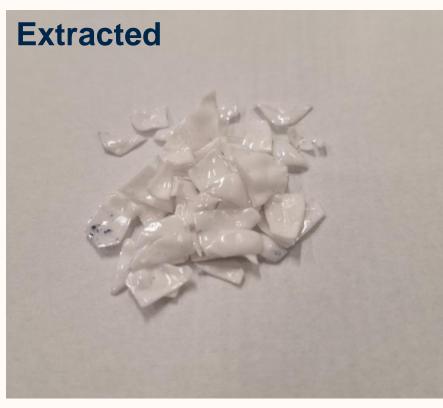
Methods



Extractor 100 mL with 5.2 L/D, flow rate 30 g/min







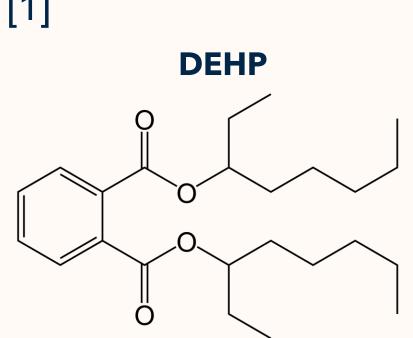
- Waste blood bag (BB) flake thickness 0.32 mm (+/- 0.01)
- Temperature and pressure study
 - 90, 110, 130 °C at 100, 200, 300 bar
 - ¹H-NMR with DMSO–d₆ (in triplo)
- Kinetic study
 - 130°C at 200 and 300 bar (in duplo)
 - ¹H-NMR with DMSO–d₆
 - Kinetic data fitting using phenomenological equation:

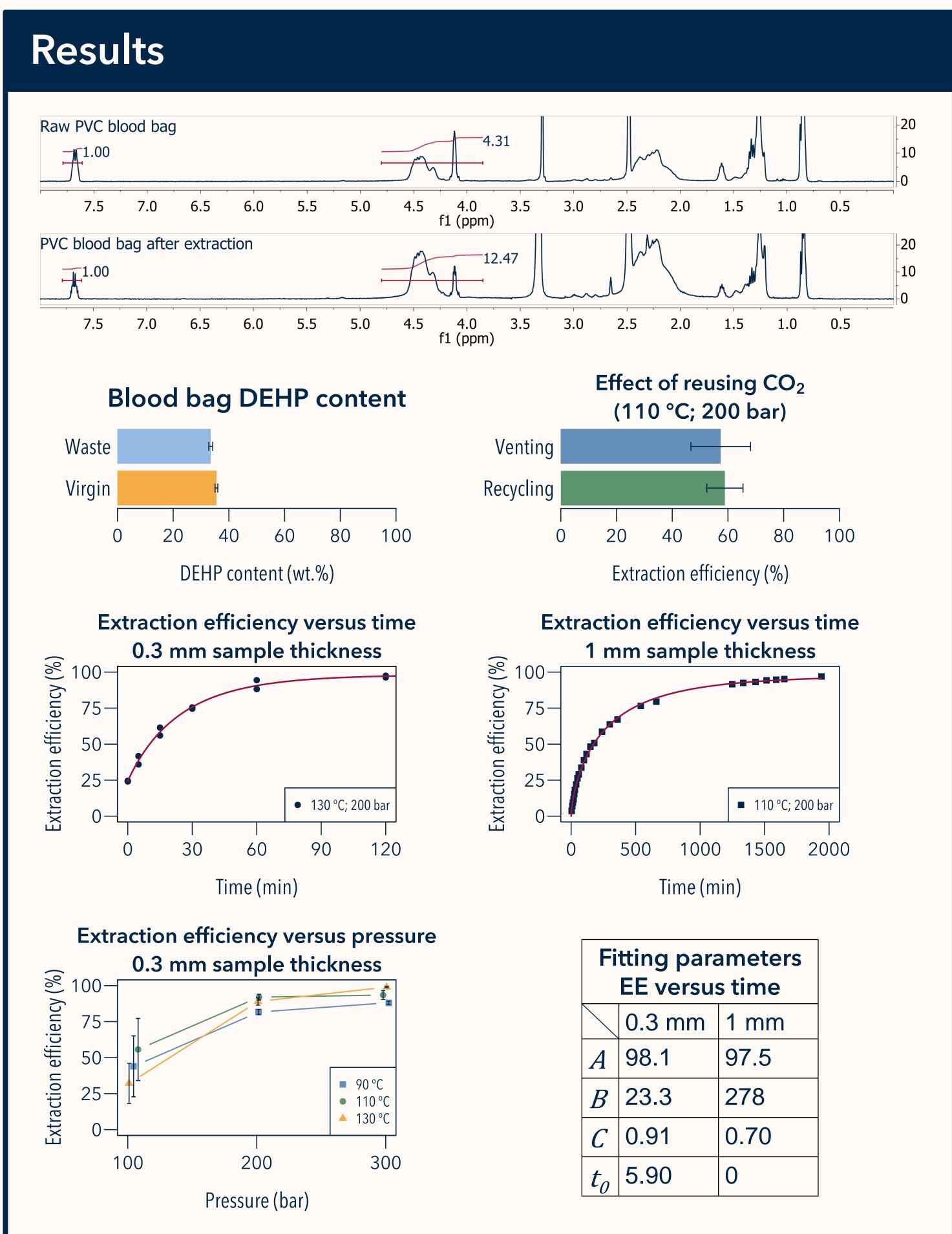
$$f(t) = A - Ae^{-\frac{(t+t_0)^C}{B}}$$

- The NMR spectra were made using MestReNova 12
- All other graphs were produced using R [4]

Introduction

- 0.1 wt.% EU phthalate limit for recycled PVC (2006) [1]
- Most flexible waste PVC still contains phthalates
 - Cables and blood bags
 - Up to 50 wt.%
- Supercritical CO₂-based extraction
 - Facile conditions: 74 bar, 31 °C [2]
 - Dissolves apolar compounds
 - Used in for example caffeine extraction [3]





Future Investigations

- Modelling of the extraction process
- Determining the solubility of DEHP in supercritical CO₂
- Understanding the effect of flow rate
- Confirming the quality of PVC after extraction
- Examining the effect of a grinding step
- Studying the rheological behavior (high pressure plate-plate)

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