# Reversibly Crosslinked Block Copolyether Micelles by Bisborane Catalyzed **Anionic Ring Opening Polymerization of Glycidyl Ethers**

#### **U. Andreas Stihl, Felix H. Schacher**

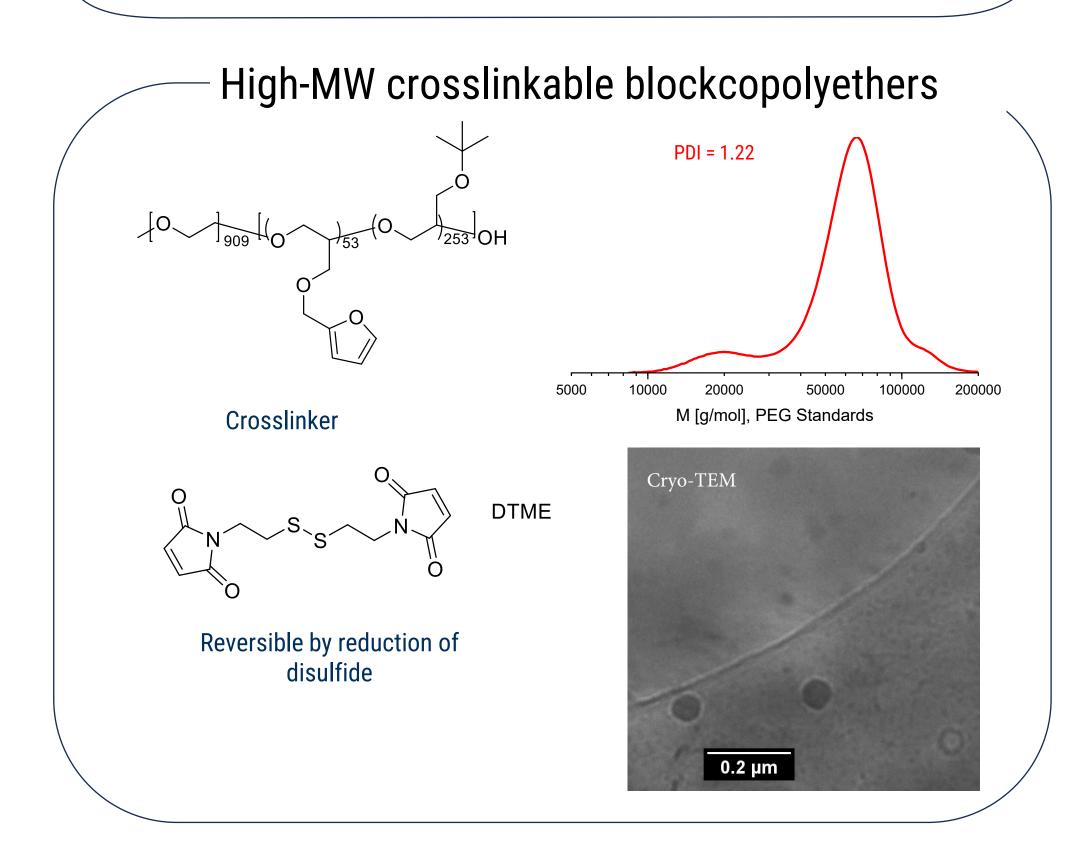
Institute of Organic and Macromolecular Chemistry (IOMC), Friedrich-Schiller-University Jena, Lessingstraße 8, D-07443 Jena and Jena Center for Soft Matter (JCSM), Friedrich-Schiller-University Jena, Philosophenweg 7, D-07743 Jena andreas.stihl@uni-jena.de felix.schacher@uni-jena.de

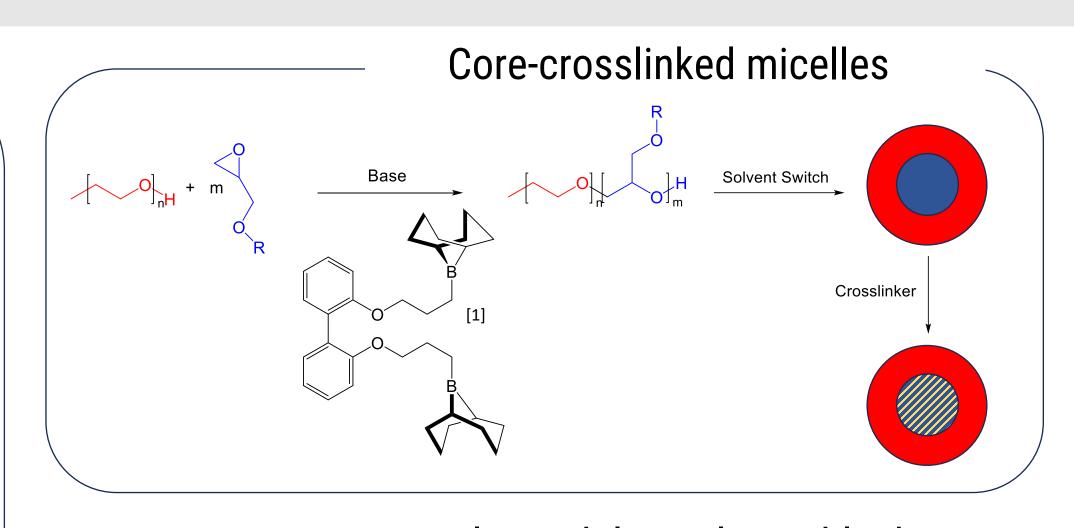
#### Motivation

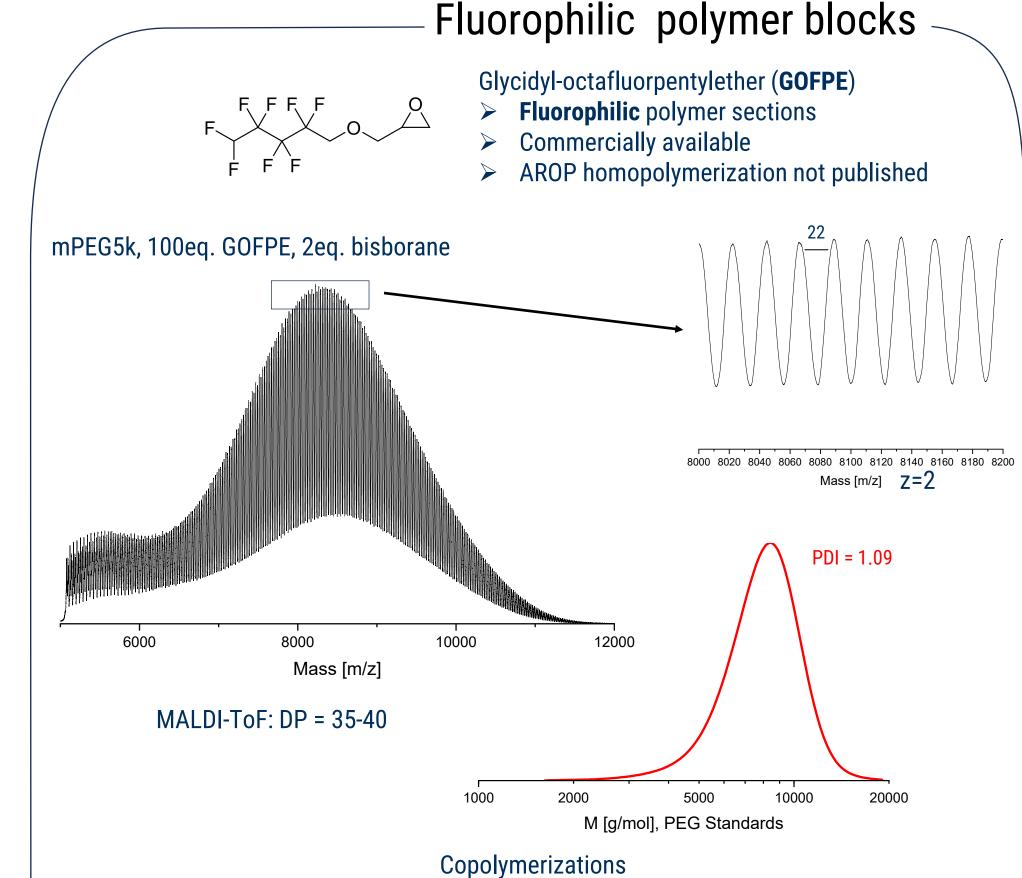
Functionalized block copolymers are a highly useful class of macromolecules due to their ability to form separated nanostructures in bulk and in solution. These make them suitable for application in surface coating, as nanoporous membranes, micellar nanocarriers or templates for mesoporous materials. Using commercially available PEGs as a macroinitiator for AROP with functionalized glycidyl ethers (GE) is a simple method of producing amphiphilic block copolymers with varying functional groups. However, if the steric or solubility characteristics of the monomer lead to reduced accessibility of the active chain end, sufficient degrees of polymerization can become difficult to achieve.

Organometallic monomer activating agents such as triethyl borane (TEB) can be used as a simple additive to increase the reactivity of GEs in AROP by withdrawing electron density from the epoxide ring by interaction with the lone electron pairs on the oxygen. However, with increasing chain lengths, the portion of TEB which interacts with the backbone instead of the monomer increases. By anchoring an additional borane moiety to the active chain end due to its negative charge, a bisborane catalyst can avoid this drawback and enable significantly higher degrees of polymerization for problematic monomers.

### pH-sensitive reversible core-crosslinking Catechol Acetonide Glycidyl Ether (CAGE) pH-dependent reversible crosslinking Surface coating via catechol adhesion Polymerization with **TEB** -> DP limited to ≈20 Cryo-TEM α-allyl-PEG5,6k α-methoxy-PEG5k PEG<sub>126</sub>-b-PCAGE<sub>11</sub> PDI=1.14 PEG<sub>114</sub>-b-PCAGE<sub>16</sub> PDI=1.07 10000 20000 50000 2000 200 nm M [g/mol], PEG Standard Polymerization with **Bisborane** -> DP exceeding 50 Cryo-TEM PEG<sub>126</sub>-b-(PCAGE<sub>60</sub>-co-tBGE<sub>6</sub>) PDI=1.12 PEG<sub>114</sub>-b-PCAGE<sub>54</sub> PDI=1.11 0.2 μm 50000 2000 5000 10000 20000 100000 M [g/mol], PEG Standard Crosslinker pH 1 **pH8** Cryo-TEM **Upon Neutralization** > crosslinking > swelling non-cl Cryo-TEM 0.2 µm $r_{H} = 26 \pm 3 \text{nm}$ $r_{H} = 32 \pm 3 nm$ 0.65 eq Crosslinker, 2mg/ml Polymer

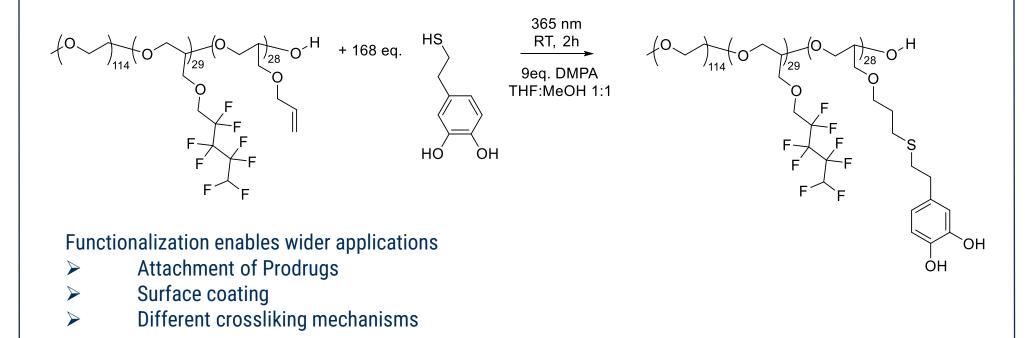






Eq. GOFPE	Comonomer	M <sub>n</sub> [kg/mol] <sup>a</sup>	Formula <sup>a</sup>	PDIb			
50	50eq. AGE	16.5	mPEG <sub>114</sub> -b-P(GOFPE <sub>29</sub> -co-AGE <sub>28</sub> )	1.14			
70	30eq. FGE	18.2	mPEG <sub>114</sub> -b-P(GOFPE <sub>39</sub> -co-FGE <sub>13</sub> )	1.17			
a) as determined by <sup>1</sup> H-NMR (300 MHz, CDCl <sub>3</sub> ), b) as determined by SEC (THF, PEG standards)							

# Functionalization by Thiol-en Click



# **Triblock copolymers**

#	Initiator	M <sub>n</sub> [kg/mol] <sup>a</sup>	Formula <sup>a</sup>	PDIb
1	APEG <sub>80</sub>	19.0	APEG <sub>80</sub> -b-PCAGE <sub>24</sub> -b-P(GOFPE <sub>26</sub> -co-FGE <sub>13</sub> )	1.18
2	APEG <sub>80</sub>	17.4	APEG <sub>80</sub> -b-PCAGE <sub>27</sub> -b-P(GOFPE <sub>19</sub> -co-FGE <sub>11</sub> )	1.11
3	mPEG <sub>114</sub>	19.2	mPEG <sub>114</sub> -b-PCAGE <sub>36</sub> -b-P(GOFPE <sub>15</sub> -co-FGE <sub>6</sub> )	1.05
4	mPEG <sub>114</sub>	16.8	mPEG <sub>114</sub> -b-P(CAGE <sub>23</sub> -co-tBGE <sub>9</sub> )-b-P(GOFPE <sub>15</sub> -co-FGE <sub>6</sub> )	1.06

a) as determined by <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>), b) as determined by SEC (THF for #1, DMAc + 0.21% LiCl for #2, CHCl<sub>3</sub>:iPrOH:NEt<sub>3</sub>

