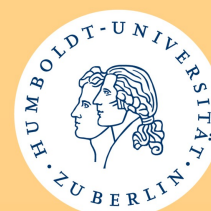


Effective Modification and Oxidative Activation of Lignin: Towards Bio-Based Thiol-Catechol 2K Adhesives

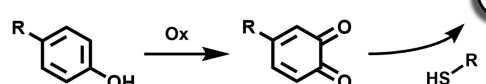
Dominik P. Hoch, Hans G. Börner*

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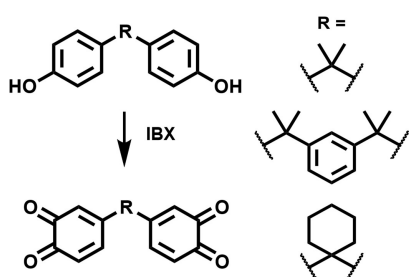


Introduction

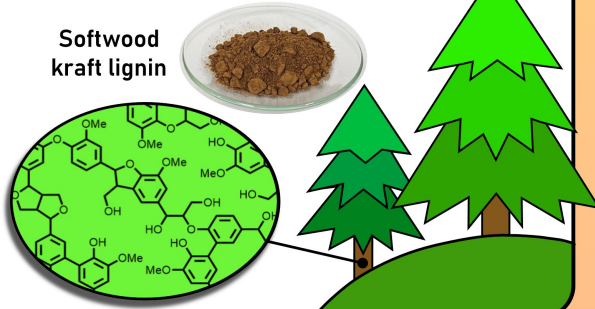
Mussel-inspired approach^[1]: chemical oxidation of phenols to *ortho*-quinones, followed by nucleophilic thiol addition to form Thiol-Catechol connectivities (TCC)



Previously established material platform^[2] based on bisphenols using IBX

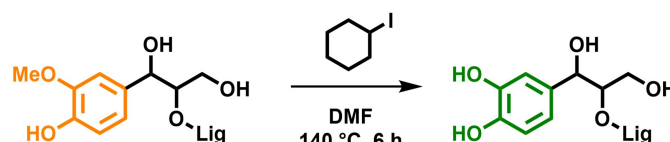


Utilizing lignin as a renewable phenol source for Thiol-Catechol Chemistry

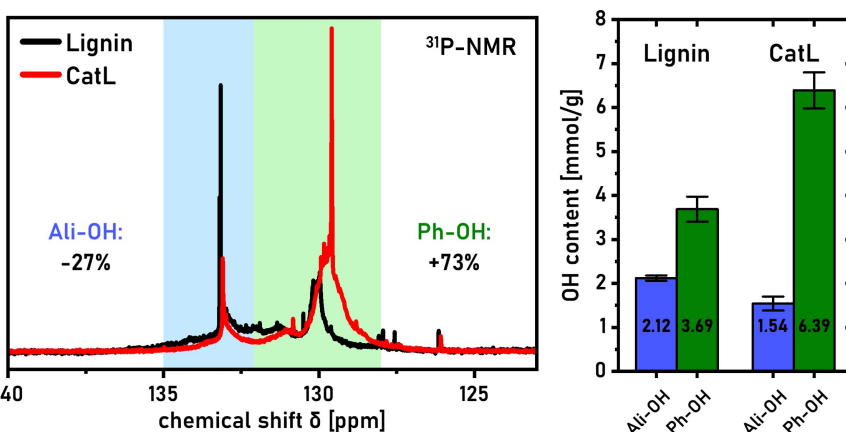
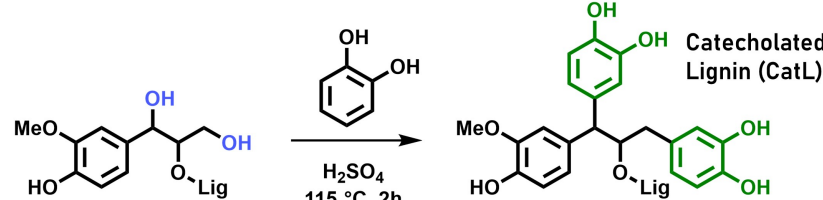


Catecholation of Lignin

Previous approach: chemical demethylation of intrinsic **methoxyphenols** (G-units) through *in situ* generated hydroiodic acid to obtain **catechols**^[3]



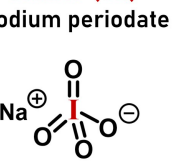
Instead: introduction of additional **catechol** functionalities through electrophilic aromatic substitution of the **aliphatic alcohols** in the Lignin structure^[4]



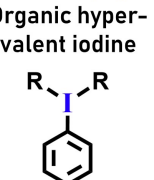
Oxidation of Lignin

-Highly reactive
-Oxidation of all OH-groups

Iodine (VII)
Sodium periodate



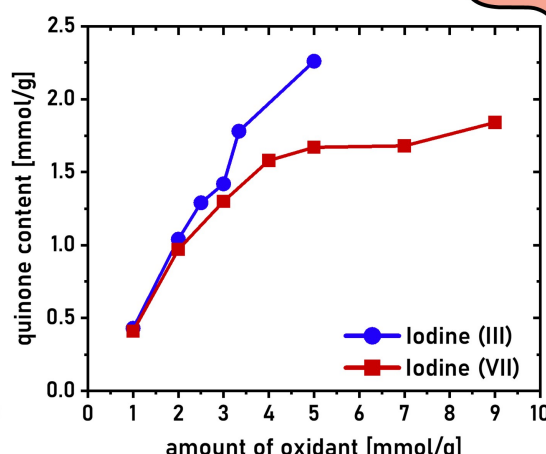
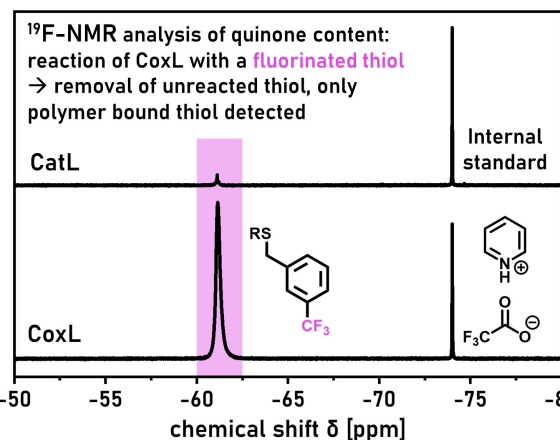
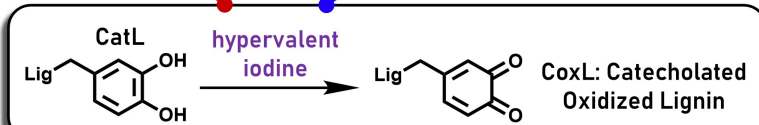
Iodine (III)
Organic hyper-valent iodine



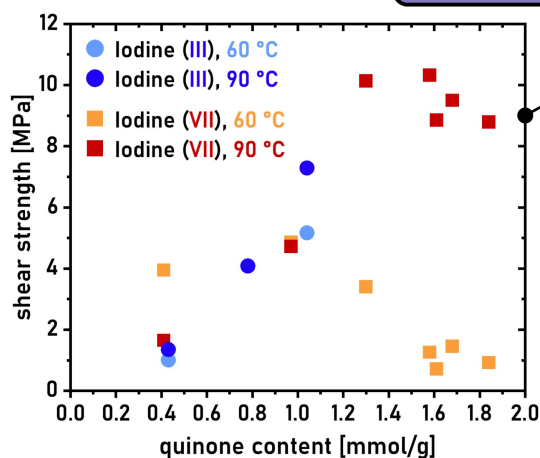
-Highly selective
-Oxidation of phenolic OH-groups

Conditions:
GVL, 10% phosphate buffer (pH=6.5)
0 °C, 1 hour

Conditions:
GVL/MeOH 2/1
RT, 15 minutes



Adhesive Performance

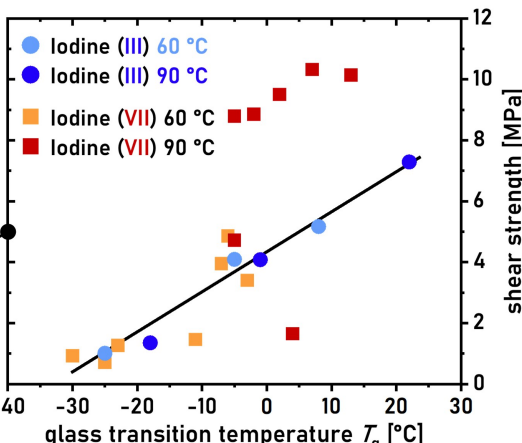


General trend of increasing shear strength with higher quinone content

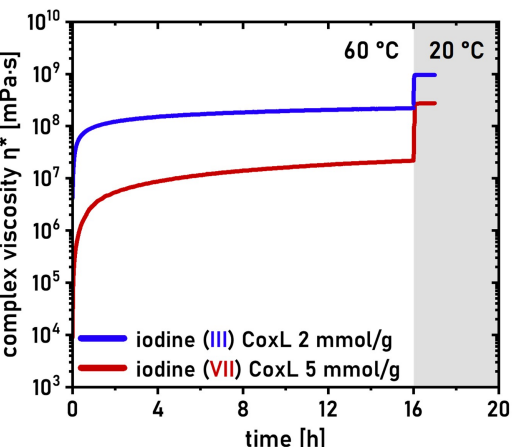
Only outlier: iodine (VII) CoxL cured at 60 °C → Incomplete curing and high degree of oxidation: lack of OH-groups for strong surface interaction

Correlation of shear strength with glass transition temperature *T_g*

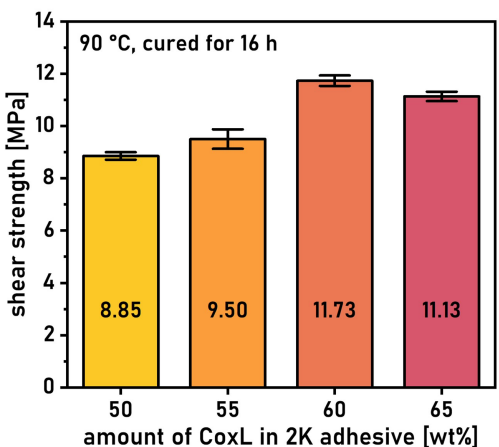
Only outlier: iodine (VII) CoxL cured at 90 °C → Lower *T_g* due to lack of OH-groups, but still high shear strength due to TCC-groups



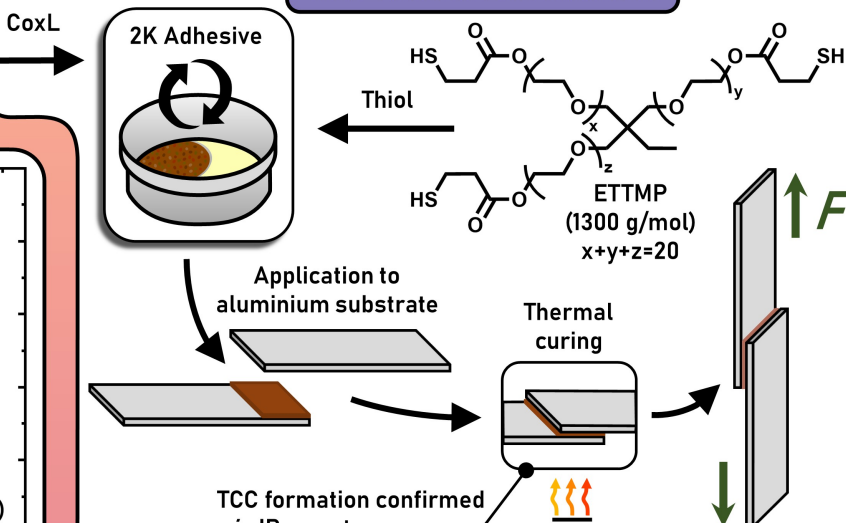
Iodine (III)-CoxL: presence of more OH-groups → better curing at lower temperatures



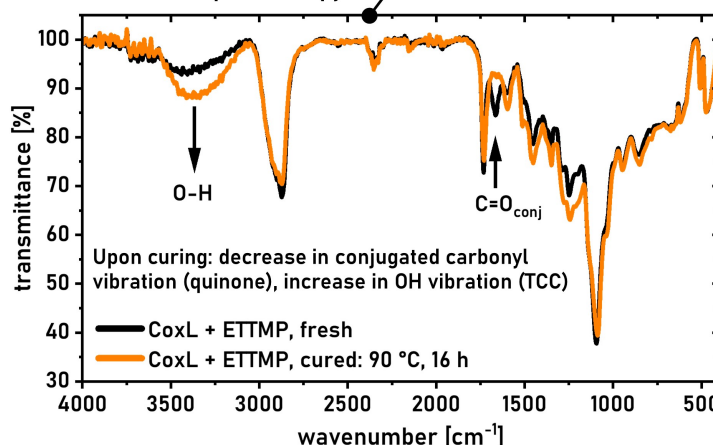
Iodine (VII)-CoxL: low viscosity mixtures → up to 65 wt% Lignin content in 2K adhesive



CoxL based 2K Adhesive



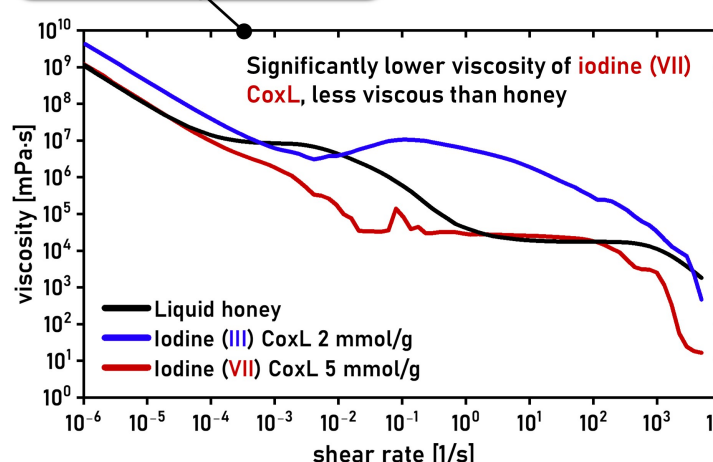
TCC formation confirmed via IR-spectroscopy



Miscibility with ETMP:

- **Iodine (III)-CoxL** only miscible up to 2 mmol/g oxidant
- **Iodine (VII)-CoxL** fully miscible

→ Selective oxidation limits miscibility, presence of large quantities of Ali-OH favors hydrogen bonding



Conclusion



Both oxidation methods have distinct advantages:

- **Iodine (III)-CoxL** offers a more selective oxidation, retaining OH-groups and thus increasing *T_g* values and performance for not fully cured systems
- **Iodine (VII)-CoxL** offers great miscibility due to the non selective oxidation of all OH-groups, allowing for strong performing adhesives with the highest amount of TCC-groups

References

- [1] J. Horsch, P. Wilke, M. Pretzler, M. Seuss, I. Melnyk, D. Remmler, A. Fery, A. Rempel, H. G. Börner, *Angew. Chem.* 2018, 130, 15954–15958.
- [2] J. M. Krüger, C.-Y. Choi, F. Lossada, P. Wang, O. Löschke, D. Auhl, H. G. Börner, *Macromolecules* 2022, 55, 989–1002.
- [3] C.-Y. Choi, F. Lossada, K. Walter, T. Fleck-Kunde, S. Behrens, T. Meinelt, J. Falkenhagen, M. Hiller, H. Oschkinat, A. Dallmann, A. Taden, H. G. Börner, *Green Chem.* 2024, 26, 2044–2058.
- [4] S. Zhao, X. Huang, A. J. Whelton, M. M. Abu-Omar, *ACS Sustain. Chem. Eng.* 2018, 6, 10628–10636.

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