Itaconate-based polyesters as potential inks for 3D printed biomaterials

Magdalena Miętus^{(1)*}, Agnieszka Gadomska-Gajadhur⁽¹⁾

(1) Faculty of Chemistry, Warsaw University of Technology, Warsaw, Poland

BACKGROUND

*magdalena.mietus.dokt@pw.edu.pl

INTRODUCTION

To date, polyacrylate and PLA-based materials are most commonly used for 3D printing.[1,2] Those materials have significant advantages (biodegradability, biocompatibility for PLA, and good mechanical properties for acylate compounds). Nonetheless, they have some major disadvantages.[3] PLA is characterized by fragility and low cell adhesion (if we consider it for use in 3D bioprinting).[2] Furthermore, acrylate compounds are toxic and are usually obtained from non-renewable raw materials, which is against the trends of Green Chemistry. Thanks to the presence of multiple bonds in the structure of acrylate polymers, they can be subjected to UV light. It makes it possible to successfully print a high-quality model of the desired shape. 3D printing can also be used to obtain cellular scaffolds with applications in tissue engineering. For this, it is necessary to use non-toxic materials that have the potential to mimic the cellular matrix naturally found in the human body.

In this research, poly(tetramethylene itaconate) (PBItc) synthesis and UV-crosslinking will be investigated. Itaconic compounds are structurally similar to acrylate ones.[4] A C=C bond in the structure of the itaconic compound allows UV-enhanced crosslinking. This work reports the findings of a research study to determine how the photocrosslinking time primarily affects the mechanical, thermomechanical, thermal, and cytotoxicity properties of the received polymer films. The aim is to identify potential applications in any of the fields of tissue engineering.

[1] 10.1021/acssuschemeng.0c02168; [2] 10.3390/polym16081140; [3] 10.1007/s10856-010-4091-8; [4] 10.1039/c6gc00605a

INFORMATION CROSSLINKING parameter Acrylic compound Itaconic compound Figure 5. Schematic representation of PBItc Figure 1. Chemical structure of acrylic acid and itaconic acid. UV-crosslinking. Advantages of itaconic acid (IA): Biodegradable Produced by fermentation (Aspergillus terreus fungi) Antibacterial and anticancer properties Presence of double bond \rightarrow post-polymerization reactions Advantages of 1,4-butanediol (1,4-BD): Biodegradable Figure 6. (A) Gel content; (B) Swelling degree of PBItc films. Naturally occurs in the human body Produced by fermentation (E. coli bacteria) PREPOLYMER SYNTHESIS Figure 7. (A) Contact angle; (B) Shore hardness of PBItc films. Figure 2. Synthesis of poly(tetramethylene itaconate) prepolymer. A) **Box-Behnken optimization** Figure 8. (A) Tensile strength; (B) Bending strength of PBItc films. > 55.0% < 52.0% < 47.0% < 42.0% < 37.0% < 32.0% A) conv_{cooH tit} → 66.1±3,9% $\%C=C_{13C \text{ NMR}} \rightarrow 35.0\%$ Figure 9. (A, B) DMTA curves for selected PBItc films. %_{VVU} **→ 95.0**% A) Figure 3. Response surface plots Figure 4. Consistency of the Figure 10. (A) Cytotoxicity effect of the 24 h extracts for selected PBItc films; of selected PBItc synthesis PBItc optimal prepolymer. (B) Degradation of PBItc (UV-cured for 4 min). parameters.

CONCLUSIONS

- ✓ PBItc was synthesized without solvent with the use of zinc acetate $Zn(OAc)_2$.
- ✓ The synthesis of PBItc was optimized and the applied statistical model fits well with the experimental results.
- ✓ UV-crosslinking time of 4 min is sufficient to obtain a polymer film with suitable properties.
- PBItc polymer films show potential to be used as a 3D printed cellular scaffold for soft tissue engineering (skin, ligament, muscle connective tissue).

Modified

Photocuring time 70 s - 15 min