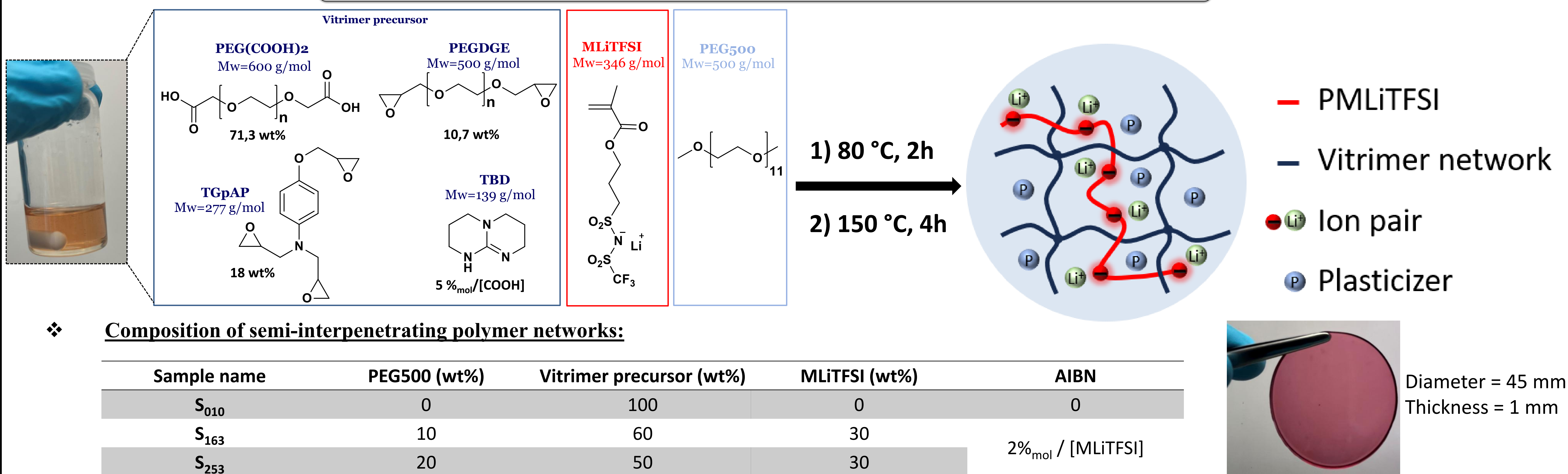


OBJECTIVE

This work aims to develop a gelified semi-interpenetrating network combining an epoxy-acid vitrimer network and an ion conducting polymer electrolyte (PLiMTFSI), with the aim of achieving improved mechanical and electrochemical performance.

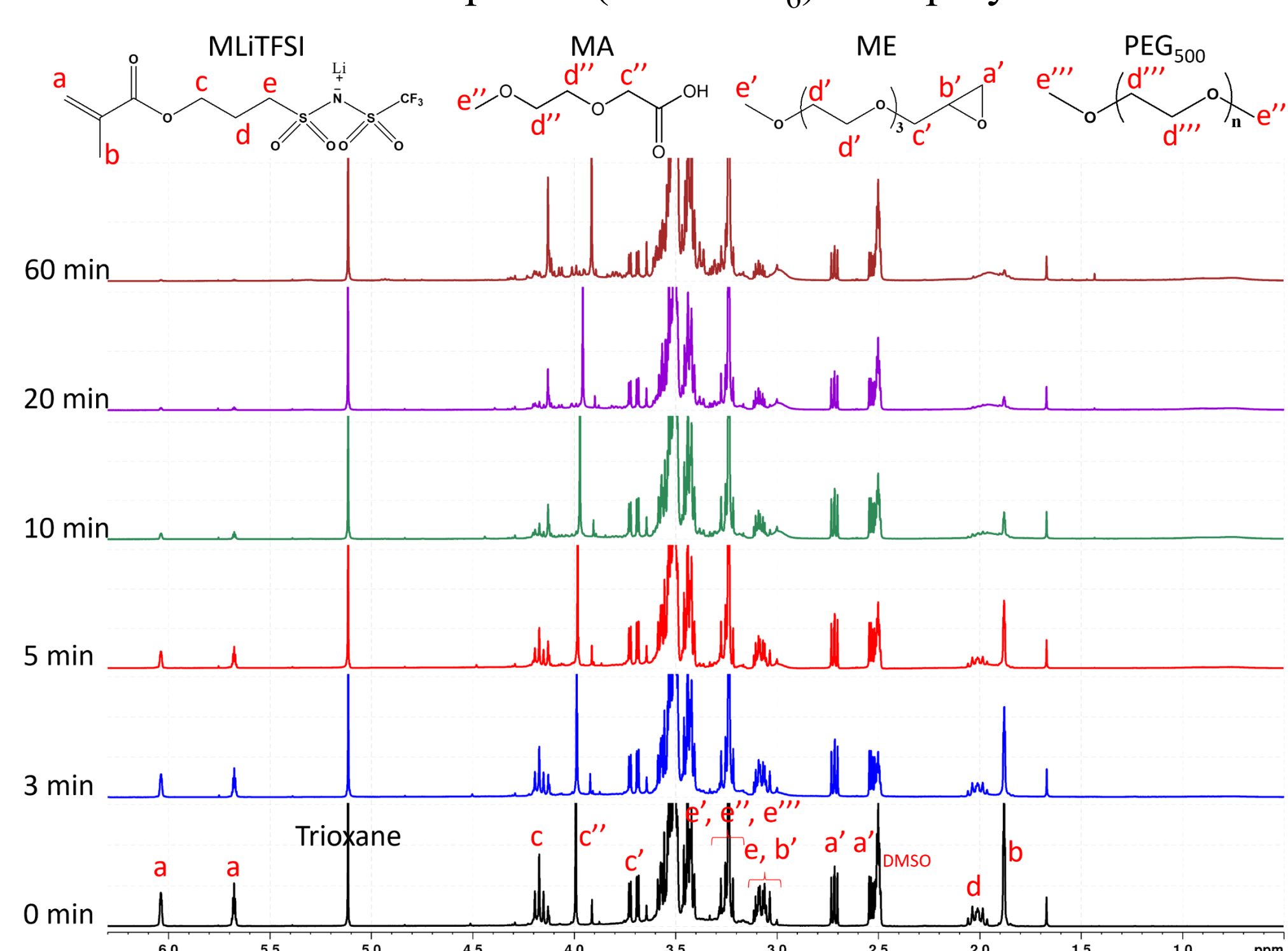
SYNTHETIC PATHWAY



RESULTS

Kinetic Study of MLiTFSI Polymerization in the Presence of Vitrimer Precursors at 80 °C

❖ Evolution of ^1H NMR spectra ($\text{DMSO}-d_6$) with polymerization time of K4

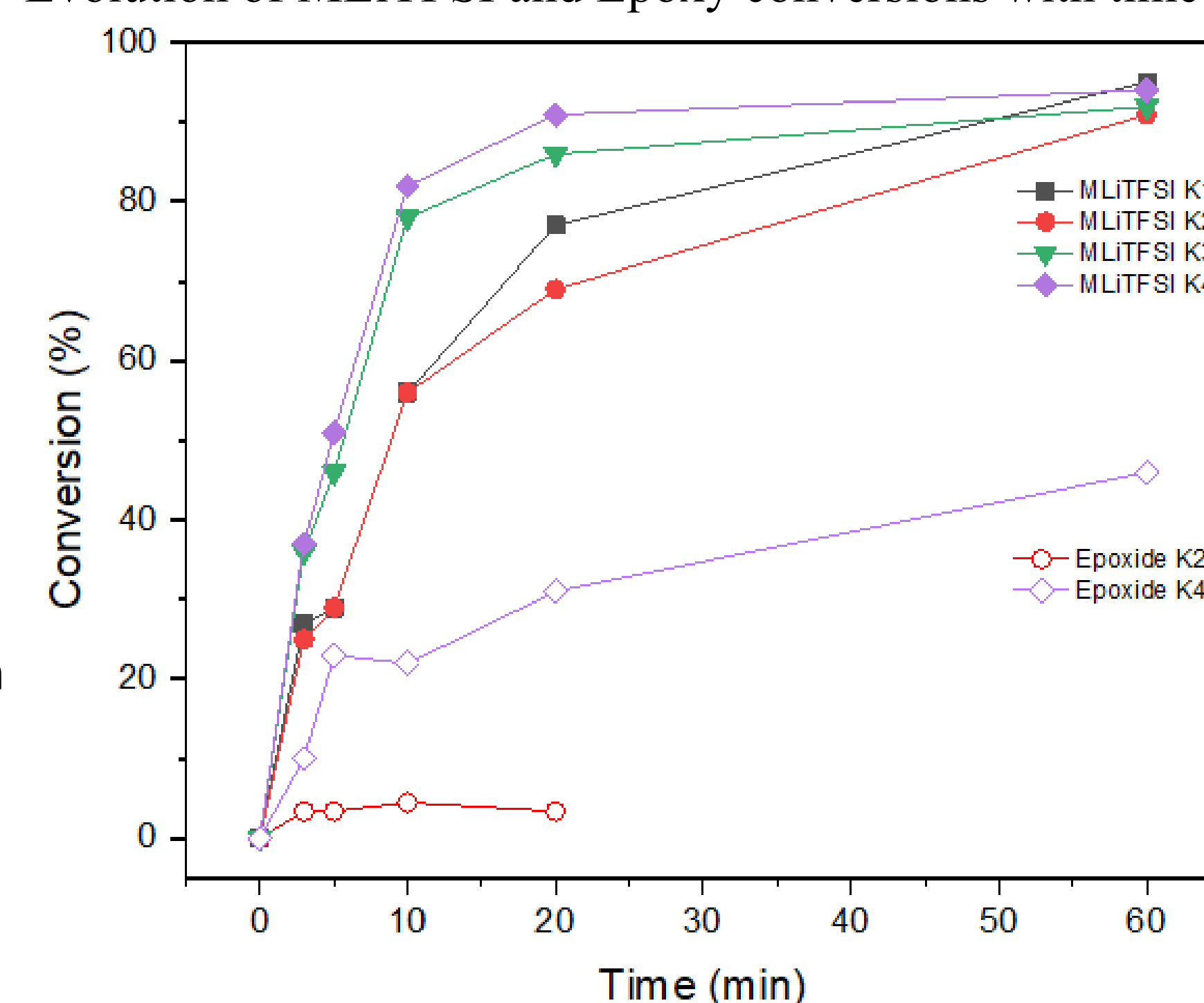


❖ Kinetic formulation of MLiTFSI polymerization

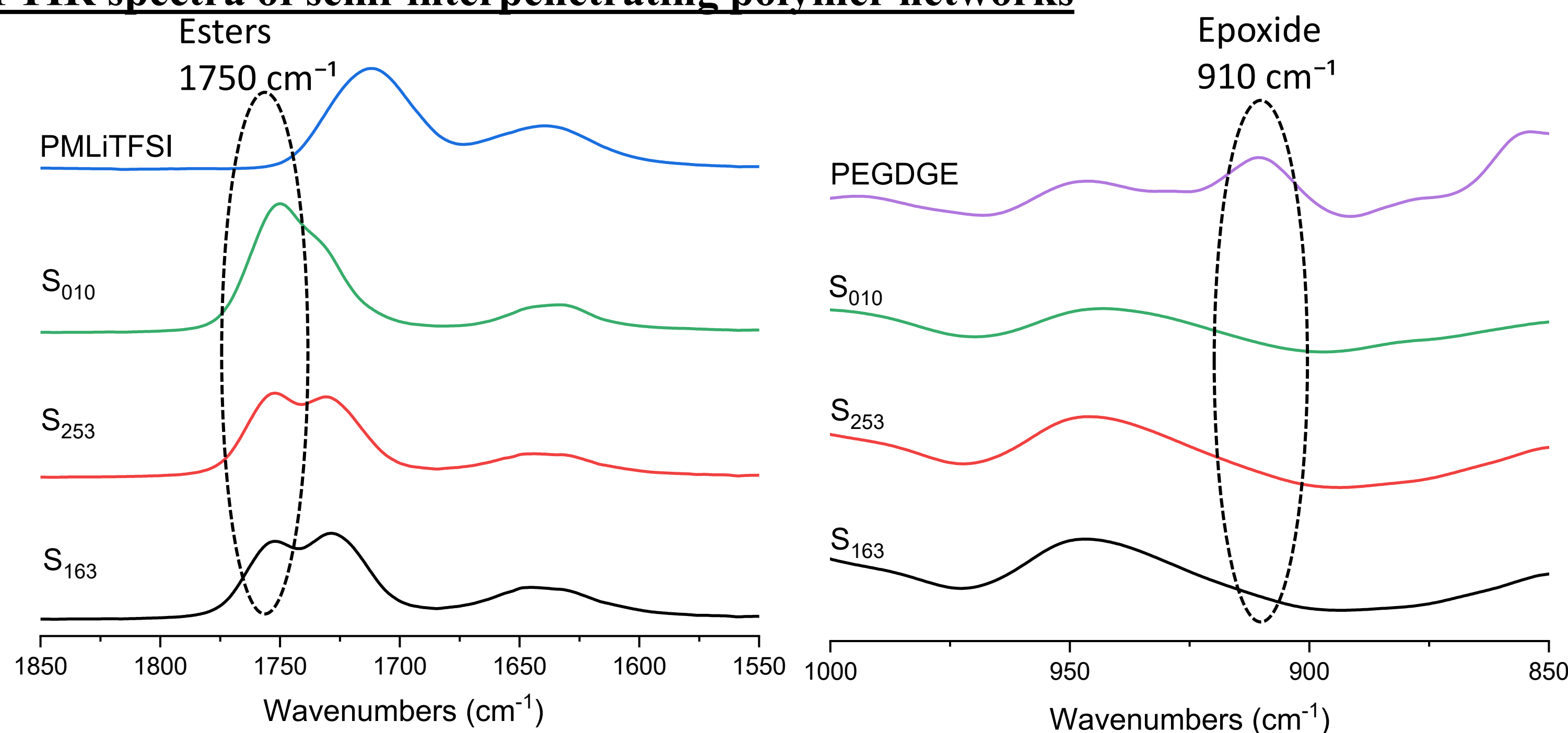
	MLiTFSi	ME	MA	AIBN
K1	[0.87]	-	-	2% _{mol}
K2	[0.87]	[1.43]	-	
K3	[0.87]	-	[1.43]	
K4	[0.87]	[1.43]	[1.43]	

- ✓ The epoxide and the acid have no influence on the polymerization of MLI⁺FSI.
- ✓ Over 90% conversion of MLI⁺FSI at 80 °C.

❖ Evolution of MLiTFSI and Epoxy conversions with time



FTIR spectra of semi-interpenetrating polymer networks



- ✓ Disappearance of epoxy at 910 cm^{-1} and carboxylic acid at 1726 cm^{-1}
- ✓ Appearance of esters at 1750 cm^{-1}

Physical and ion conducting properties

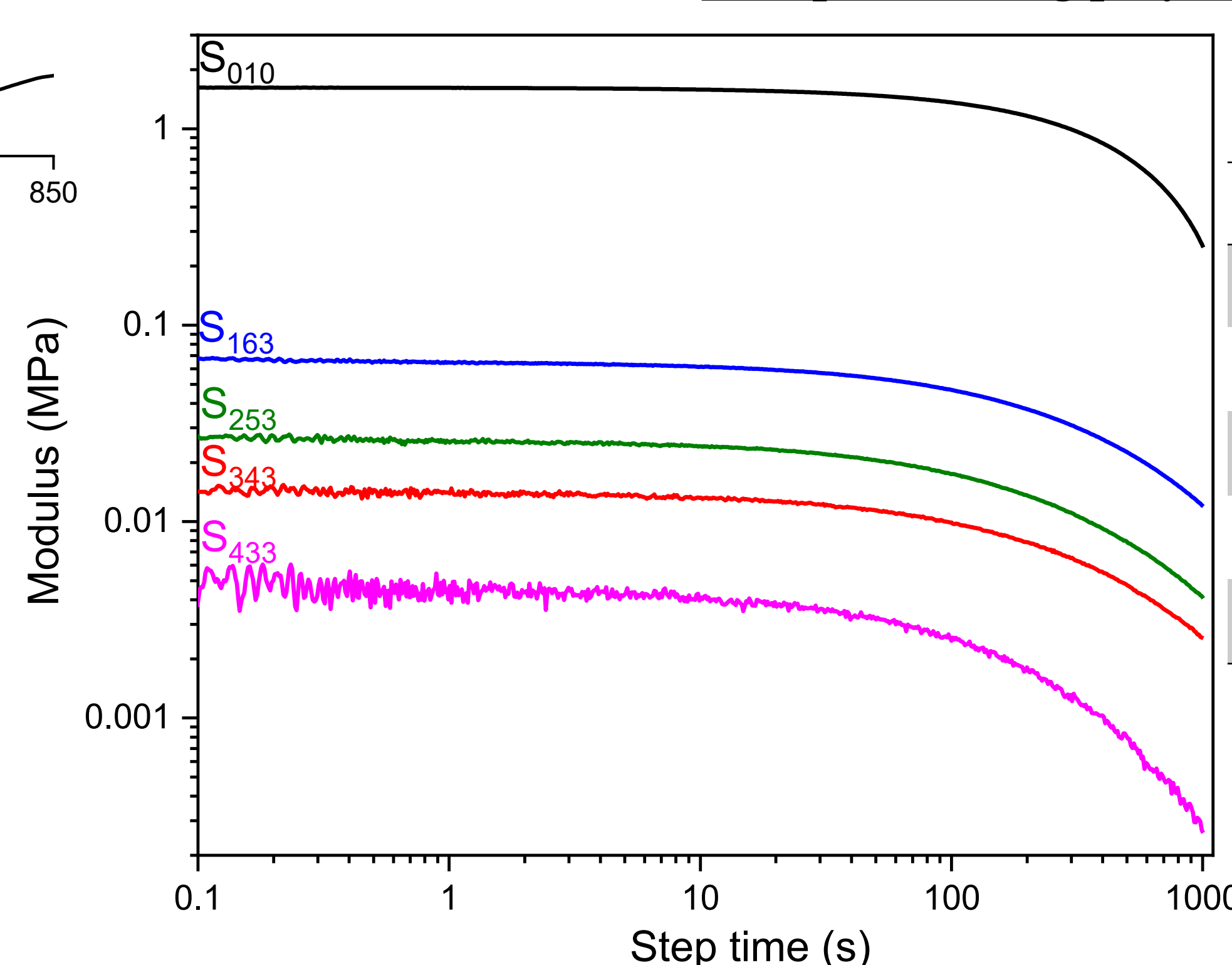
	S_{010}	S_{163}	S_{253}	S_{343}	S_{433}
T_g [°C]	- 37	- 18	- 22	- 30	- 48
G_N [MPa]	1.60	0.06	0.02	0.01	0.004
σ [S/Cm]	-	1.1×10^{-5}	1.9×10^{-5}	3.3×10^{-5}	4.5×10^{-5}

Swelling Properties of semi-interpenetrating polymer networks in THF

	S_{010}	S_{163}	S_{253}	S_{343}	S_{433}
Gel content (%)	95	97	96	99	99
Swelling ratio	2.1	1.7	2.0	1.4	1.7

- The plasticizer is the only component considered extractable in the solvent

Rheology monitoring at 180 °C of the stress relaxation of semi-interpenetrating polymer networks



	$t_{\text{gel}} \text{ (s)}$	$t_{\text{relaxation}} \text{ (s)}$
S_{010}	600	180
S_{163}	440	220
S_{253}	350	250
S_{343}	420	420
S_{433}	200	500

CONCLUSION

In summary, a dynamic gelified semi-interpenetrating network was successfully developed by combining an epoxy-acid vitrimer with the single-ion conducting polymer PLiMTFSI. FTIR, swelling, and rheology confirmed the formation and dynamic behavior of the network, while the resulting materials exhibited ionic conductivity on the order of 10^{-5} S/cm at 60 °C, comparable to values reported for analogous single-ion systems