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Aim of the work	Materials								
<ul style="list-style-type: none"> Incorporation of organic Phase Change Materials (PCMs) into a thermoplastic matrix for 3D printing of structures for passive cooling of electronic devices Maximize melting enthalpy of the composite material Enhance thermal conductivity 	<ul style="list-style-type: none"> PCM → SA stearic–palmitic acid mixture Matrix → HDPE: High Density Polyethylene “Filler” → EG: Expanded Graphite 								
Sample preparation									
1) First shape stabilization	<p>[1] Sacchet, S.; Valentini, F.; Benin, A.; Guidolin, M.; Po, R.; Fambri, L. Expanded Graphite (EG) Stabilization of Stearic and Palmitic Acid Mixture for Thermal Management of Photovoltaic Cells. <i>C</i> 2024, 10, 46. https://doi.org/10.3390/c10020046</p> <p>[2] Sacchet, S.; Valentini, F.; Guidolin, M.; Po, R.; Fambri, L. Shape-Stabilized Phase Change Materials with Expanded Graphite for Thermal Management of Photovoltaic Cells: Selection of Materials and Preparation of Panels. <i>Appl. Sci.</i> 2025, 15, 4352. https://doi.org/10.3390/app15084352</p>								
2) Second shape stabilization	<p>[3] Sacchet, S.; Valentini, F.; Rizzo, C.; Po, R.; Fambri, L. High density polyethylene with phase change materials for thermal energy management. <i>Energy Mater.</i> 2025, 5, 500042. http://dx.doi.org/10.20517/energymater.2024.1112</p> <p>Nominal composition</p> <table border="1"> <thead> <tr> <th>Sample</th> <th>HDPE (wt.%)</th> <th>SA (wt.%)</th> <th>EG (wt.%)</th> </tr> </thead> <tbody> <tr> <td>PE-50SA/EG14</td> <td>50</td> <td>43.9</td> <td>6.1</td> </tr> </tbody> </table>	Sample	HDPE (wt.%)	SA (wt.%)	EG (wt.%)	PE-50SA/EG14	50	43.9	6.1
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PE-50SA/EG14	50	43.9	6.1						
Results	<p>Thermogravimetric analysis (TGA)</p> <p>36.7 wt.% residual mass at 700°C (SA), 56.9 wt.% (PE-50SA/EG14), 6.4 wt.% (HDPE).</p> <p>Differential Scanning Calorimetry (DSC)</p> <p>Melting enthalpy: 62 J/g (PE-50SA/EG14)</p> <p>Good starting point, but the goal is usually >100 J/g (i.e. change PCM)</p> <p>Microscopy Images</p>								
CONTAINED LEAKAGE	<p>Bulk density</p> <p>Coefficient of Thermal Expansion (CTE):</p> <ul style="list-style-type: none"> Solid: 8.2×10^{-4} Liquid: 6.9×10^{-4} <p>For the PCM: volume measurements + IR thermography</p>								
Thermophysical properties	<p>Thermal Conductivity enhancement</p> <p>$\lambda = \alpha \cdot \rho \cdot c_p$</p> <ul style="list-style-type: none"> λ: thermal conductivity ($\frac{W}{m \cdot K}$) α: thermal diffusivity ($\frac{mm^2}{s}$) ρ: bulk density ($\frac{g}{cm^3}$) c_p: specific heat capacity ($\frac{J}{g \cdot K}$) 								
Three-point bending test	<p>ASTM D790-17 standard</p> <p>No broken specimens</p> <p>Dynamic Mechanical Analysis (DMA)</p> <ul style="list-style-type: none"> Thermal inertia effects Superposition of HDPE β transition and PCM melting 								
Future perspectives	<ul style="list-style-type: none"> Optimization of the composition Use of other organic PCMs (goal of 100 J/g) Optimize the process to obtain a more homogeneous filament <p>APPLICATION:</p> <ul style="list-style-type: none"> Thermal management of electronic devices Complex geometries Design finned systems to enhance heat dissipation (PCM recrystallization) 								